



## Potential Energy Surfaces (PES)

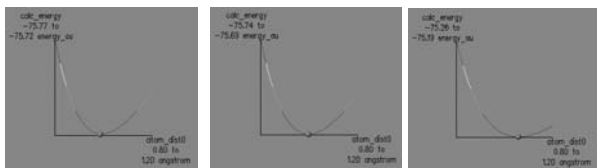
$E$  and other properties are function of geometry

- Mathematical relationship of  $E$  as a function of structure
- Surface has as many dimensions as number of internal degrees of freedom in molecule
  - ▶ Diatomic molecule AB
    - $r_{AB}$
  - ▶ Nonlinear triatomic molecule ABC
    - $r_{AB}$  and  $r_{BC}$  (unless identical)
    - A-B-C bond angle

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## Diatomic Molecule PES

Two-dimensional Potential Energy Well with Bond Length as Variable



$r(\text{OH}^-) = 0.99 \text{ \AA}$   
(0.9628)

$r(\text{OH}) = 1.00 \text{ \AA}$   
(0.9706)

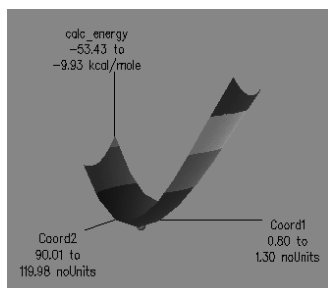
$r(\text{OH}^+) = 1.06 \text{ \AA}$   
(1.0289)

B88-LYP/DZVP

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## Triatomic Molecule PES

Surface with Bond Length and Bond Angle as Variables



$r(\text{OH}) = 0.950 \text{ \AA}$   
(0.958)

H-O-H =  $107.7^\circ$   
( $104.5^\circ$ )

PM3

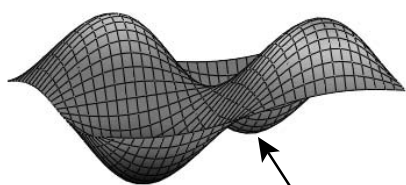
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## PES Features

### ■ Minima

- ▶ Bottom of "Valley" on PES
  - Changing any geometric parameter increases  $E$
- ▶ Equilibrium structures of molecule
  - Different conformers
  - Structural isomers
  - Reactant and product

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Local minimum:  
(lowest point in a  
limited region of PES)

Global minimum:  
(lowest point on PES)

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### ■ Maxima

- ▶ Top of "mountain" on PES
  - Changing any geometric parameter decreases  $E$

### ■ Saddle Point

- ▶ Maximum for one molecular parameter and minimum for another
  - Transition structure between two equilibrium structures

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- Advantages of SD
  - ▶ Always locate (local) minimum
- Problems of SD
  - ▶  $\mathbf{g}_{i+1}$  is perpendicular to  $\mathbf{g}_i$ 
    - Lose any further  $E$  lowering possible by  $\mathbf{g}_i$
    - Path oscillates around minimum path
  - ▶ Rate of convergence decreases near minimum
  - ▶ Undesireable behavior in long, narrow valley

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- Modifications of SD
  - ▶ Fit 3 points to function
    - Differentiate to find minimum
  - ▶ Arbitrary step size
    - Start with predetermined value or value proportional to  $\mathbf{g}_i$
    - Calculate  $E$  at new position
    - If  $E$  is less
      - Increase step size
      - Calculate  $E$  at new position
      - Repeat until  $E$  increases
    - If  $E$  is greater or increases
      - Decrease step size
      - Calculate  $E$  at new position
      - Iterate

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## Optimization Methods

Conjugate Gradient Method (CG)

- Mixture of current gradient and previous search direction
- Procedure
  - ▶ First step: SD
  - ▶ Subsequent steps:  $\mathbf{v}_{i+1} = -\mathbf{g}_{i+1} + \gamma_{i+1} \mathbf{v}_i$ 
    - Fletcher-Reeves:  $\gamma_{i+1} = (\mathbf{g}_{i+1} \cdot \mathbf{g}_{i+1}) / (\mathbf{g}_i \cdot \mathbf{g}_i)$
    - Polak-Ribiere (common):  $\gamma_{i+1} = [(\mathbf{g}_{i+1} - \mathbf{g}_i) \cdot \mathbf{g}_{i+1}] / (\mathbf{g}_i \cdot \mathbf{g}_i)$
    - Hestenes-Stiefel:  $\gamma_{i+1} = [(\mathbf{g}_{i+1} - \mathbf{g}_i) \cdot \mathbf{g}_{i+1}] / [\mathbf{v}_i \cdot (\mathbf{g}_{i+1} - \mathbf{g}_i)]$

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## Global Minimum

- Optimization Results
  - ▶ Above methods locate the “nearest” minimum (local?)
  - ▶ Conformers and transition structures are related to local minima and saddle points
  - ▶ Molecular properties are related to global minimum--the most stable structure
  - ▶ Several approaches to finding the global minimum
    - NO Guarantees!!!

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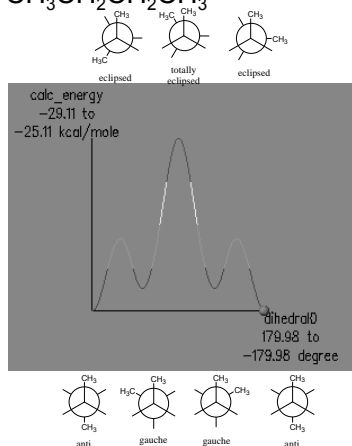
## Global Minimum Methods

### Grid Search

- Simple
  - ▶ Calculate all possible energies
  - ▶ Prepare energy map or table
  - ▶ Choose best structure
- Problems
  - ▶ Number of minima increases exponentially with the number of variables
  - ▶ Not practical for large or biomolecules
    - Build structures from optimized fragments
    - Optimize portions of molecule

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Example:  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$



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Example:  $\text{CH}_3(\text{CH}_2)_{n+1}\text{CH}_3$

$n = 1$	$N = 3^1 = 3$	$t = 3 \text{ s}$
$n = 2$	$N = 3^2 = 9$	$t = 9 \text{ s}$
$n = 5$	$N = 3^5 = 243$	$t = \sim 4 \text{ min}$
$n = 10$	$N = 3^{10} = 59 \text{ k}$	$t = \sim 16 \text{ h}$
$n = 15$	$N = 3^{15} = 14 \text{ M}$	$t = \sim 160 \text{ d}$

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## Global Minimum Methods

Other

- Monte Carlo
  - ▶ Usually start at a minimum
  - ▶ Randomly change one or several angles or bond lengths to generate new geometry
  - ▶ Calculate  $E$ 
    - If  $E$  is lower, accept new geometry
    - If  $E$  is greater, retain or reject based on a Boltzmann distribution related to the energy change and  $T$
  - ▶ Many variations

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- Molecular Dynamics
  - ▶ Based on Newton's equations of motion for atoms
  - ▶ Molecule overcomes barrier between minima if kinetic energy is great enough
    - Similar to climbing out a valley over a mountain
  - ▶ Kinetic energy is proportional to  $T$ 
    - Use elevated temperatures (600 - 1200 K)
    - High temperatures reduce chance of trapping molecule in a local minimum
  - ▶ Essentially searches PES in the region of the starting minimum

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